

## 物理化學二-量子化學 (PHYSICAL CHEMISTRY II-Quantum

### Chemistry)

(課號 203 33170)

#### **學分：**

三學分 (一學期，大三上學期開課)

#### **預修課程：**

普通化學、化學數學或相等科目

#### **開授對象：**

化學系大三學生之基礎必修課。

課程內容：	授課時數
<b>一、量子理論(Quantum theory) :</b> 光電效應與輻射的粒子性(Photoelectron effect and particlelike properties of radiation) 粒子的波動性(Wavelike properties of particles) 海生堡測不準原理(The Heisenberg uncertainty principle) 德布洛意物質波(De Broglie matter wave) 薛丁格方程(The Schrodinger equation) 算子(Operators) 量子力學的基本假設(Basic postulates of quantum mechanics)	<b>6 h</b>
<b>二、粒子在一維盒子的運動(Particle in a box) :</b> 一維盒子中的粒子(Particle in a one-dimensional box) 應用到共軛多烯(Application to conjugated polyenes) 粒子在三維盒子的運動(Particle in a three-dimensional box)	<b>2 h</b>
<b>三、諧振子(The harmonic oscillator) :</b> 古典諧振子(The classical harmonic oscillator) 量子諧振子(The quantum harmonic oscillator)	<b>4 h</b>
<b>四、氫原子(Hydrogen atom) :</b> 類氫原子的薛丁格方程(The Schrödinger equation for hydrogenlike atoms) 類氫原子的本徵函數與機率密度(Eigenfunctions and probability density for hydrogenlike atoms) 類氫原子的軌域角動量(Orbital angular momentum of hydrogenlike atoms) 角動量(Angular momentum) 自旋(Spin)	<b>4 h</b>
<b>五、估計方法(Approximation methods) :</b> Rayleigh-Schrödinger 微擾理論 (Rayleigh-Schrödinger perturbation theory) 變分法(Variation method)	<b>6 h</b>
<b>六、多電子原子的電子結構(The electronic structure of many-electron atoms) :</b> 氦原子(Helium atom) 包立不共容原理(Pauli-exclusion principle) 氦的第一激發態(First excited of helium atom) 鋰原子(Lithium atom)	<b>6 h</b>

週期表與建構原理(The periodic table and the Aufbau principle)

原子的項符號(Atomic term symbols)

**七、分子的電子結構**(The electronic structure of molecules) :

**6 h**

Born-Oppenheimer 近似 (The Born-Oppenheimer approximation)

氫分子離子(The hydrogen molecule ion)

氫分子的分子軌域描述(The molecular orbital description of the hydrogen molecule)

同核雙原子分子的電子組織(Electron configurations of homonuclear diatomic molecules)

**八、分子與輻射的作用**(The interaction between molecules and radiation) :

**4 h**

電磁輻射(Electromagnetic radiation)

光的吸收與發射(Absorption and emission of light)

躍遷機率(Transitional probabilities)

選擇律(Selection rules)

**九、光譜**(Spectroscopy) :

**6 h**

旋轉與振動光譜(Rotational and vibrational spectroscopy)

核運動的薛丁格方程(Schrödinger equation for nuclear motion)

雙原子分子的旋轉光譜(Rotational spectra of diatomic molecules)

雙原子分子的振動光譜(Vibrational spectra of diatomic molecules)

雙原子分子的振動旋轉光譜(Vibration-rotation spectra of diatomic molecules)

多原子分子的振動光譜(Vibrational spectra of polyatomic molecules)

拉曼光譜(Raman spectra)

電子光譜(Electronic spectroscopy)

電子能級與選擇律(Electronic energy levels and selection rules)

Franck-Condon 原理(The Franck-Condon principle)

振子強度(Oscillator strength)

多原子分子的電子光譜(Electronic spectra of polyatomic molecules)

共軛分子：自由電子模型(Conjugated molecules: Free-electron model)

螢光與磷光(Fluorescence and phosphorescence)

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合計 **44 h**